

ADDITION REACTIONS OF CYCLIC FATTY ACID EPOXIDES

Gj.Stefanović, I.Pejković-Tadić, B.Terecki and

B.Jakovljević-Simonović

Institute of Chemistry, Faculty of Science, Studentski trg 16

Belgrade, Yugoslavia

(Received 16 September 1966; in revised form 24 October 1966)

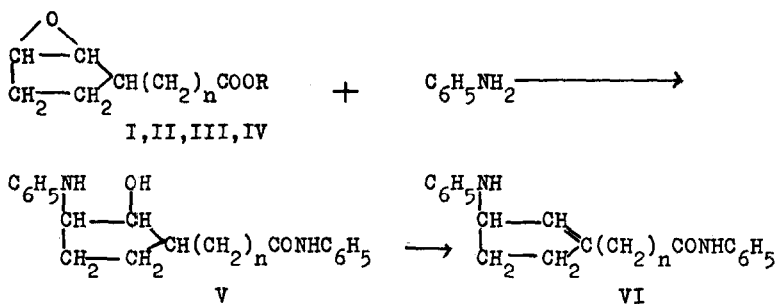
Examples of amine addition to higher fatty acid epoxides are rare (1,2), and addition to higher cyclic fatty acids is unknown. We have found that aniline, hydrazine and ammonia add to epoxides of methyl hydnocarpate and chaulmoograte forming definite crystalline products.

Hydnocarpic and chaulmoogric acids were allowed to react with ethereal perphthalic acid for 24 hours. The resulting epoxides (I, m.p. 62°; II, m.p. 72°; IR^{*}, epoxide bands at 1250, 925, 840 cm⁻¹ and carbonyl at 1700 cm⁻¹) were esterified (diazomethane) and addition reactions carried out with their esters (III, m.p. 28°; IV, m.p. 32°; IR, ester carbonyl at 1725 cm⁻¹).

Reaction of IV with an excess of aniline at 220-230° (8 hours) under nitrogen atmosphere afforded a crystalline product melting at 120° (methylene chloride-petroleumether). On the basis of elemental analysis, IR spectrum (NH at 3250 cm⁻¹, amide bands at 1650, 1545, 1325 cm⁻¹, mono-substituted phenyl at 1600, 1495, 1440, 758 and 690 cm⁻¹), mol. wt. determination by the Barger-Rast method (Found: 444; Calc.: 446) the product was recognized as the derivative

* All IR spectra were taken as KBr discs.

of higher amino acid (VI). The reaction occurred with opening of the oxirane ring, followed by the dehydration of the intermediate hydroxy amino compound (V):



$n = 10$ (I, III), 12 (II, IV, V, VI)

$R = \text{H}$ (I, II), CH_3 (III, IV)

Treatment of epoxy chaulmoograte (IV) in a sealed tube for 8 hours either with anhydrous hydrazine ($220-230^\circ$) or with liquid ammonia (100°), gave in good yields white crystallizable products which exhibited great tendency to polymerize. Elemental analyses, mol. wt. determinations and IR data of these new compounds could be accommodated by the formulae of a tetramer ($\text{C}_{72}\text{H}_{136}\text{O}_6\text{N}_{10}$) and a dimer ($\text{C}_{36}\text{H}_{65}\text{O}_5\text{N}$) respectively. Their structures are now under examination.

A full account of this and related work will be given at a later date.

ACKNOWLEDGEMENT. We are grateful for support of this work to the Research Fund of PR Serbia, Grant No.1005 of 9/X/1964.

REFERENCES

1. G. V. Figulevskii and I. L. Kuranova, J. Gen. Chem. U.S.S.R., 24, 2006 (1954).
2. H. P. Kaufmann, G. Hauschild and R. Schickel, Fette, Seifen, Anstrichmittel, 63, 246 (1961).